

COMMON POLYATOMIC IONS

| NAME | FORMULA |
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| Ammonium | NH_4^+ (the only common polyatomic cation) |
| Acetate | $\text{C}_2\text{H}_3\text{O}_2^-$ (or CH_3COO^-) |
| Carbonate | CO_3^{2-} |
| Hydrogen carbonate (bicarbonate) | HCO_3^- |
| Hypochlorite | ClO^- |
| Chlorite | ClO_2^- |
| Chlorate | ClO_3^- |
| Perchlorate | ClO_4^- |
| Chromate | CrO_4^{2-} |
| Dichromate | $\text{Cr}_2\text{O}_7^{2-}$ |
| Nitrate | NO_3^- |
| Nitrite | NO_2^- |
| Permanganate | MnO_4^- |
| Phosphate | PO_4^{3-} |
| Hydrogen phosphate | HPO_4^{2-} |
| Dihydrogen phosphate | H_2PO_4^- |
| Sulfate | SO_4^{2-} |
| Hydrogen sulfate (bisulfate) | HSO_4^- |
| Sulfite | SO_3^{2-} |
| Hydrogen sulfite (bisulfite) | HSO_3^- |
| Hydroxide | OH^- |
| Cyanide | CN^- |
| Peroxide | O_2^{2-} |